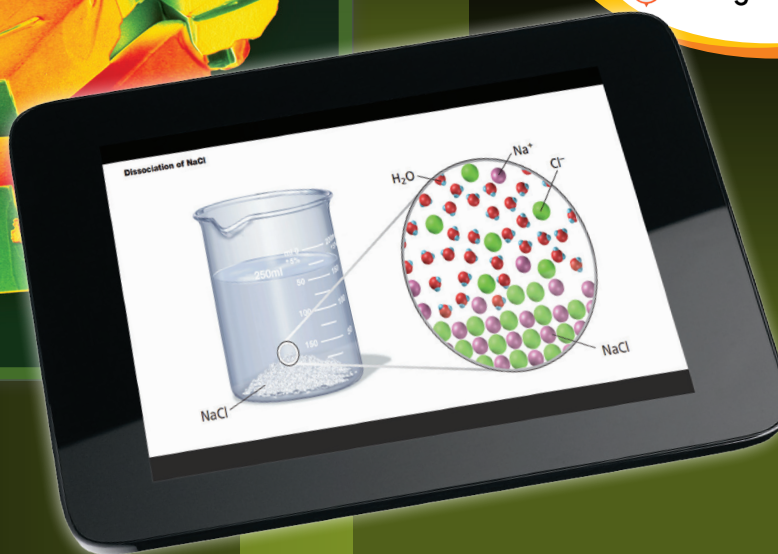
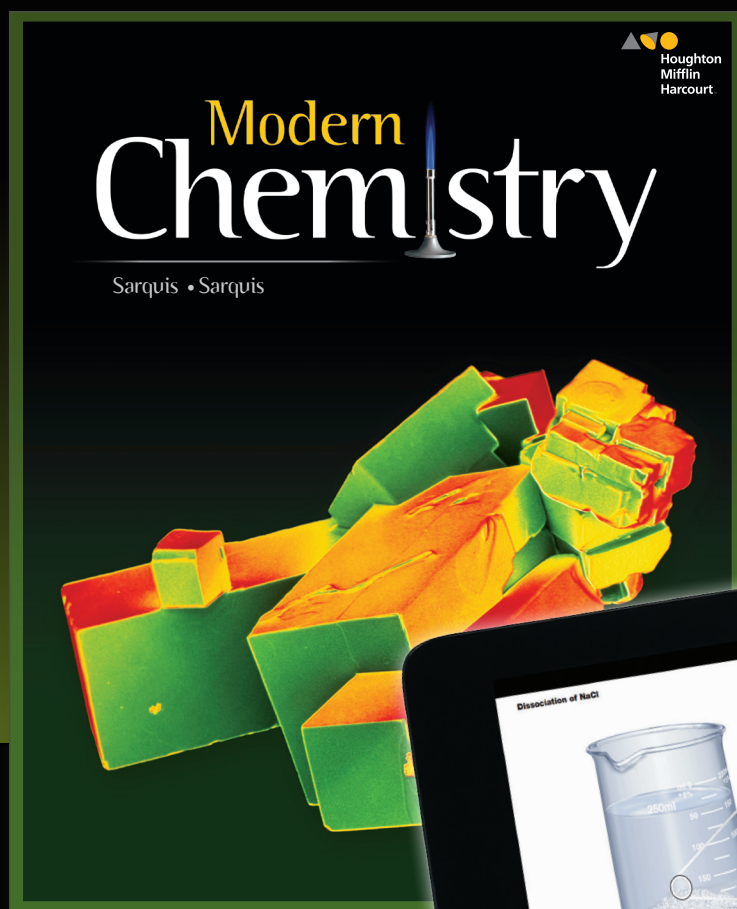


Modern Chemistry

© 2017 | REVIEWER'S GUIDE



Featuring content
from **THING**
EXPLAINER and
 Google Expeditions

Houghton Mifflin Harcourt

Modern Chemistry[®]

Less paper, more **convenience**

Everything you need—now in one convenient online location!

The Interactive Online Edition gives students and teachers 24/7 point-of-use access to all program components.



Dashboard

Classrooms using **Modern Chemistry** © 2017 will now have the benefit of the **improved** online interface provided by the HMH Dashboard. This also includes *mySmartPlanner*, enabling teachers to combine calendar functionality with curriculum mapping and program resources.

Try it now!

Just follow these steps to see how interactive and engaging online resources can be!

1	Go to:	HMHScience.com
2	Click on	PREVIEW
3	Enter Sample Word and Click Next:	HSNASC17
4	Fill in the Required Personal Information, Click the Checkbox to Agree to the Terms of Use and Privacy Policy, and Click:	Register
5	Write Down Your User Name and Password and Log in at:	HMHScience.com

Any Device, Anytime, Anywhere

Why It Matters

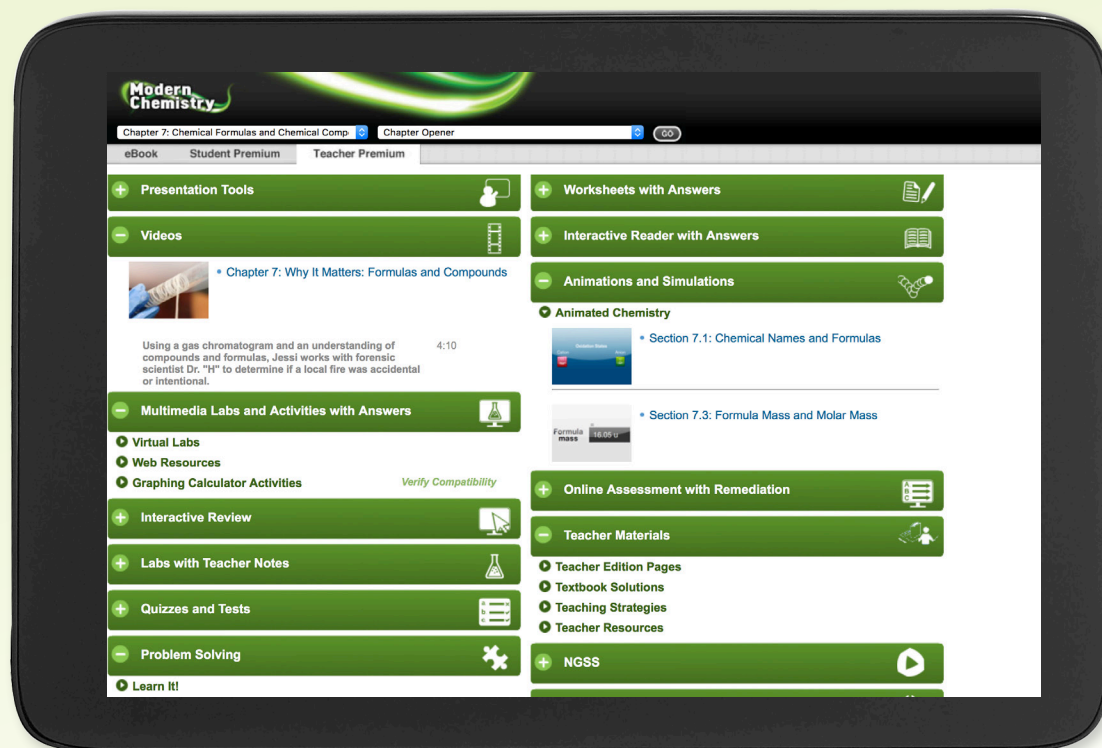
Each chapter opens with a dynamic video that relates the content to everyday life.

Virtual Labs

Students can conduct meaningful experiments in a simulated lab or field setting without the expense, time, or risk of traditional lab settings.

Animated Chemistry

Animations and simulations help students visualize and comprehend complex concepts.

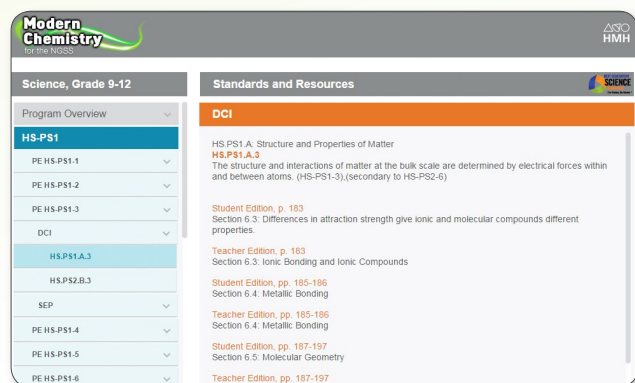


Learn It! Videos

Tutorial videos walk students through challenging problems and offer tips for success.

Solve It! Cards

Portable reference cards offer quick access to strategies for solving almost any chemistry problem.



NGSS* Correlations

Correlations both online and in the TE facilitate standards implementation.



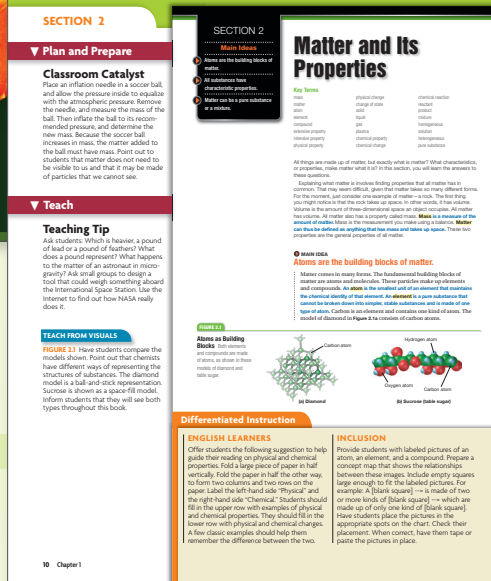
Print components **designed** and **aligned** for easy access

HMH Modern Chemistry enables you to reach all learners by providing time-saving, easy-to-use resources to help students of all abilities achieve understanding and success.



Student Edition

Offers features that make chemistry concepts more accessible, such as **highlighted vocabulary, problem-solving support**, and references to online student support tools.



Teacher Edition

Packed with a wide variety of **strategies** to help all students master chemistry concepts, plus **extended learning** opportunities for advanced students.



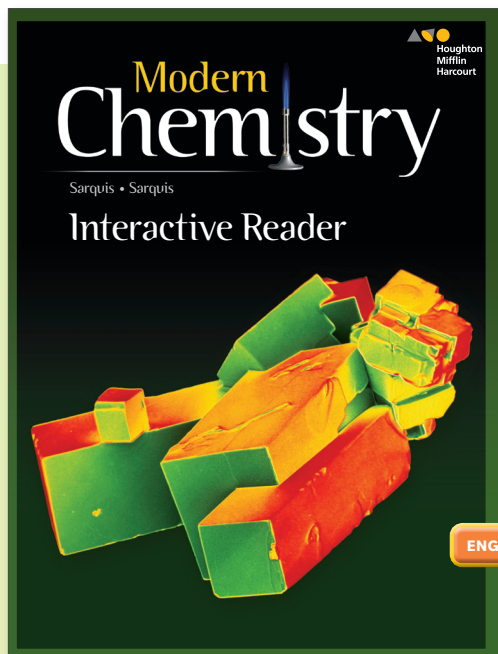
Dr. Jerry Sarquis,
Professor Emeritus,
Chemistry Education,
Miami University



Mickey Sarquis,
Professor Emerita,
Chemistry Education,
Miami University

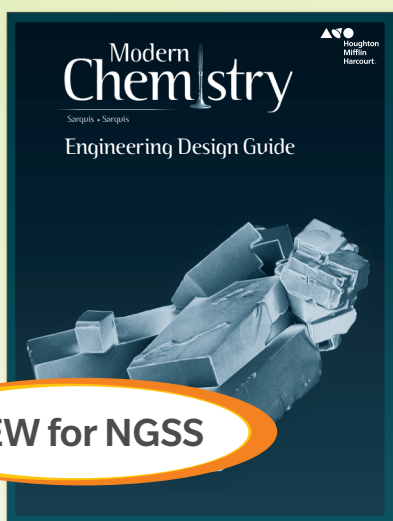
Meet the Authors

Dr. Jerry Sarquis and **Mickey Sarquis** were both professors in the Department of Chemistry at Miami University in Oxford, Ohio. These renowned authors were motivated to contribute to **Modern Chemistry** because they had a desire to give back to the community, and they wanted to influence students whom they couldn't even see, through the unfolding of a textbook and all the resources that support such a book. Jerry and Mickey Sarquis got into teaching because of their love of learning, and the more they taught, the more they learned from their interactions with their students.



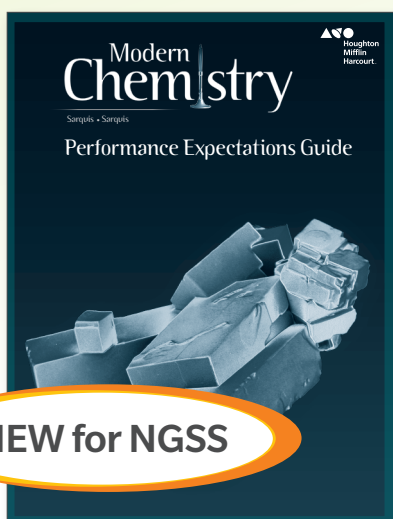
Interactive Reader and Answer Key

A **write-in worktext** that provides all of the essential content and vocabulary of the Student Edition at a reading level one to two grades below the text. A great resource for students of all ability levels, the Interactive Reader is both a core instructional tool for **struggling students** and a useful **study guide** for all students. The Answer Key provides teacher notes and answers for every section of the Interactive Reader.



Engineering Design Guide Student Edition and Teacher Edition

This Engineering Design Guide provides an overview of the **engineering design process**, along with activities and checklists that can help foster students' **critical-thinking** and **problem-solving skills**. For curriculums aligned to NGSS*, this guide can also help support the engineering-related Performance Expectations.



Performance Expectations Guide Student Edition and Teacher Edition

Designed to integrate easily into any curriculum, a separate Performance Expectations Guide is available to ensure that students meet the NGSS Performance Expectations. Also included is an **overview of NGSS** and **teacher tips** for integrating each activity into the classroom.

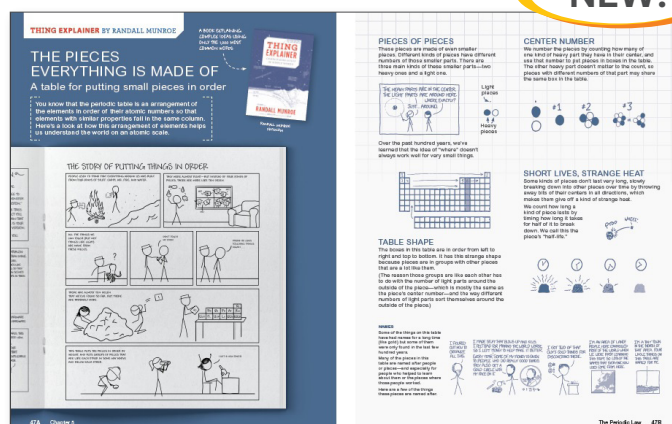


Print and Digital Tools That Motivate and Engage

HMH Modern Chemistry offers the latest print and multimedia resources that speak directly to your students in a visual language they understand—ensuring that they will stay engaged.

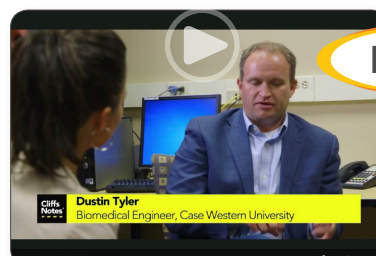
Thing Explainer

Through an exclusive partnership with author and Internet sensation **Randall Munroe**, HMH has incorporated highly engaging and educational material from Randall's latest book, *Thing Explainer*, into our print and digital editions. Randall's webcomic style, as seen on **xkcd.com**, humorously explains complex topics in easy-to-understand language.



Google Expeditions

Through its alliance with Google®, HMH is developing content for Google® Expeditions. Using a simple Google Cardboard™ device and a smartphone, students are swept away into **immersive virtual worlds** where learning and engagement are maximized. These virtual field trips are 3D, 360-degree experiences in fascinating locations, directly tied to content! A **Teacher Guide** provides ideas for incorporating the Expeditions into your lessons, as well as tips on **how to guide and customize the experience**.

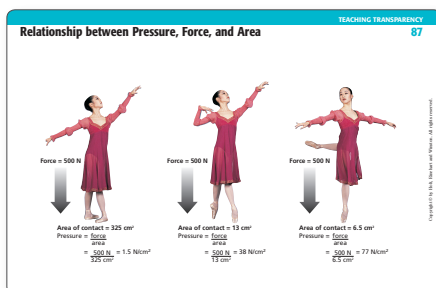


On the Job STEM Videos

As part of our Premium offering, HMH now includes 29 *On the Job* STEM videos that **profile STEM careers** in today's fastest-growing industries. Our energetic hosts shadow passionate professionals in a day "on the job." These short segments are inspirational and entertaining with the hosts actually performing parts of the job! These videos will **motivate students** to enter emerging STEM fields.

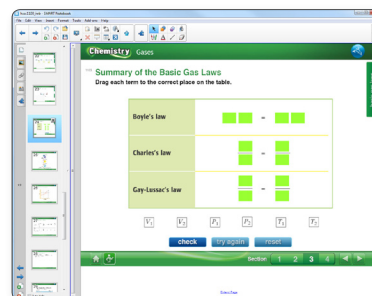
Why It Matters Videos

Seventeen chapter-introductory videos are an effective way to begin a new topic of chemistry study. Each video is designed to take the content of the chapter and relate it to everyday objects or situations that are familiar to students.



Teaching Visuals

Digital versions of key illustrations and diagrams are ideal for **whole-class instruction**.



Interactive Whiteboard Resources

IWB resources include interactive **teaching visuals** and **content-reinforcement lessons** for each chapter of the textbook.



Animated Chemistry

Each Animated Chemistry includes 3 parts—an overview of the concept, an interactive simulation, and assessment.

Unparalleled resources for Differentiated Instruction

Students approach chemistry with a wide variety of skills and levels of preparation. **HHM Modern Chemistry** gives teachers what they need to help all students succeed.

CHAPTER 11 Instruction and Intervention Support Find all of your resources online at HMDScience.com.

1 Core Instruction

2 Support and Intervention

3 Specialized Support

400A Chapter 11

Teacher Edition

The **Instruction and Intervention** feature located in each chapter of the Teacher Edition provides **strategies for every lesson** to assist you in helping students with a wide range of needs. To simplify lesson planning, these support pages at the beginning of each chapter provide a full listing of the activities and classroom resources available for each section.

The wrap margin includes a **Differentiated Instruction** feature with a wide variety of strategies to help all students master chemistry concepts. Material categories include Below Level, English Learners, Pre-AP®, and Inclusion.

Differentiated Instruction

BELOW LEVEL

Have students do a jigsaw activity to introduce this chapter. Divide the class into groups of 3–4 students and give each group a number. Divide Section 1 of this chapter into 3 or 4 parts, and name them A, B, C, etc. Give each student in a group a letter designation in addition to their group number. For example, a student might be designated as 1A. Have the A students from all groups discuss and determine the major concepts within part A of the section. The B students will do the same with part B, and so on.

Allow about 15 minutes for discussion. Then, have students report back to their original, numbered group, teaching the group what their designated part of the section had to say.

Chapter and Section Study Guide

The student worksheets in this guide cover the content in each section of the textbook using a **variety of questioning strategies**.

Editable!

NAME: _____ CLASS: _____ DATE: _____

CHAPTER 11 REVIEW

Gases

MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided.

1. Consider the following data table:

Altitude above sea level	Atmospheric pressure (kPa)
0	101.3
1000	90.0
2000	80.0
3000	70.0
4000	60.0
5000	50.0

a. Explain briefly why the pressure decreases as the altitude increases.

b. A few places on Earth are below sea level (the Dead Sea would be one such place). What would be true about the average atmospheric pressure there?

2. Explain how the ideal gas law can be simplified to give the equation $P \propto T$, when the pressure and temperature are constant.

PROBLEMS Write the answer on the line to the left. Show your work.

3. Convert a pressure of 0.800 atm to the following units:

a. torr _____

b. Pa _____

NAME: _____ CLASS: _____ DATE: _____

CHAPTER 11 REVIEW

Gases

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

1. Pressure is defined as $\frac{\text{force}}{\text{area}}$. For a constant force, when the surface area is tripled the pressure is _____

(a) doubled
(b) a third as much
(c) tripled
(d) unchanged

2. Rank the following pressures in increasing order:

(a) 50 kPa
(b) 2 atm
(c) 76 torr
(d) 100 N/m²

3. Explain how to calculate the partial pressure of a dry gas that is collected over water when the total pressure is atmospheric pressure.

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

4. a. Use five to six data points from Appendix Table A.8 in the text to sketch the curve for water vapor's partial pressure versus temperature on the graph provided below.

b. Do the data points lie on a straight line? _____

c. Based on your sketch, predict the approximate partial pressure for water at 11°C _____

Interactive Reader Audio Files

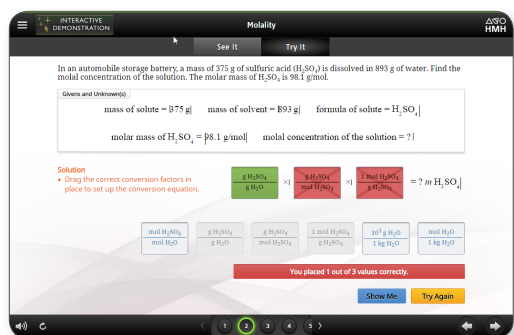
The entire Interactive Reader has been professionally read and is available to students to help **bolster learning comprehension**.



and Problem Solving

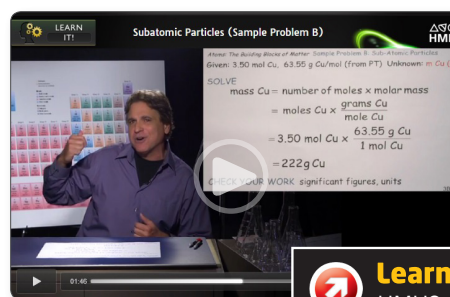
Nearly half of the sample problems in **HM Modern Chemistry** have been refreshed to give even the most loyal program users something new and different to challenge and strengthen their students' problem-solving skills.

Solution Tutor
Guides students step-by-step through selected problems, recognizes their error patterns, then provides hints and targeted remediation to improve their problem-solving skills.



Interactive Demonstrations

Each sample problem in the textbook has an accompanying Interactive Demonstration that **walks through the steps of solving** that type of chemistry problem. The Try It Yourself feature helps students apply what they have learned. Each includes a full audio narrative.



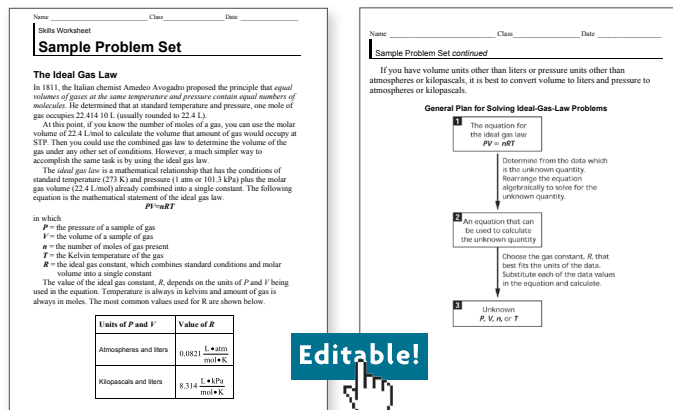
Learn It! Video
HMHSscience.com

Learn It! Videos

Forty professional tutorial videos **walk students through challenging chemistry problems**, with tips and strategies for success.

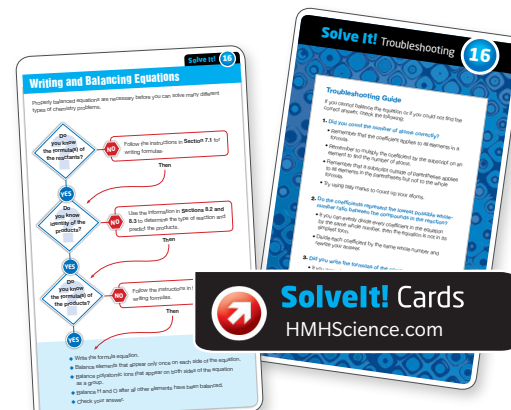
Sample Problem Sets

These skills worksheets provide **problem-solving strategies** and an extensive bank of student **practice problems** for every type of chemistry problem in the textbook.



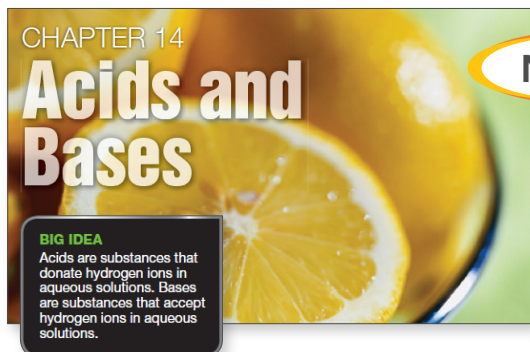
Solve It! Cards

These printable and **portable reference cards** provide students with quick access to effective problem-solving strategies and guidelines.



Wide-ranging support for Reading and Vocabulary

Your students will get the most out of their reading with numerous student and teacher print and multimedia point-of-use resources that enable them to build understanding and retain more information on key concepts.



NEW!

Student Edition

Big Ideas in every Chapter Opener & Summary help students concentrate on key concepts.

Main Idea

Chapter content has been organized around main ideas.

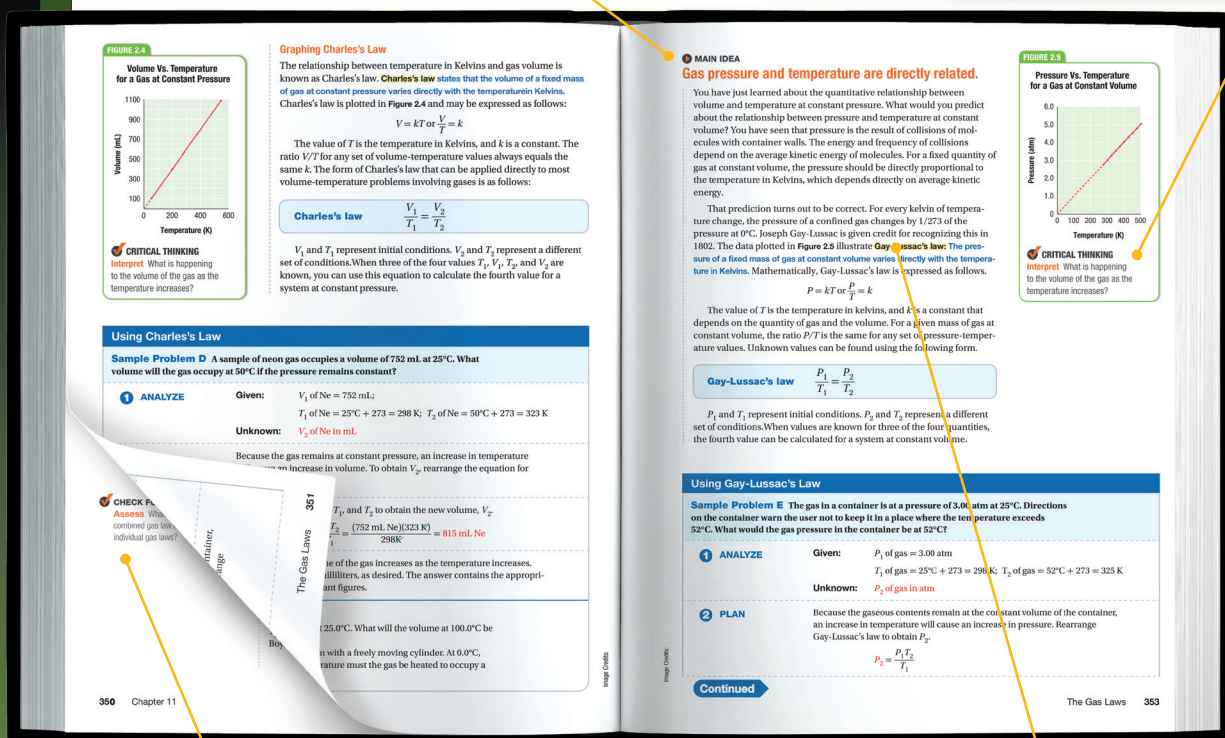
Critical Thinking

Critical-Thinking questions prompt your students to think deeply.



CRITICAL THINKING

Interpret What is happening to the volume of the gas as the temperature increases?



CHECK FOR UNDERSTANDING

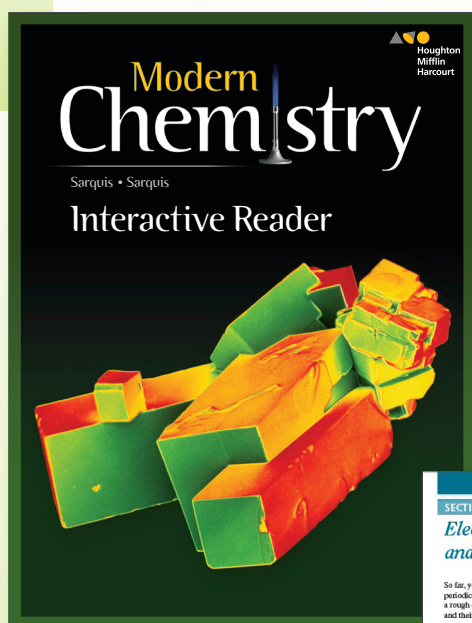
Assess What advantage does the combined gas law have over the three individual gas laws?

Check for Understanding

These reading comprehension questions help reinforce the important points of the section.

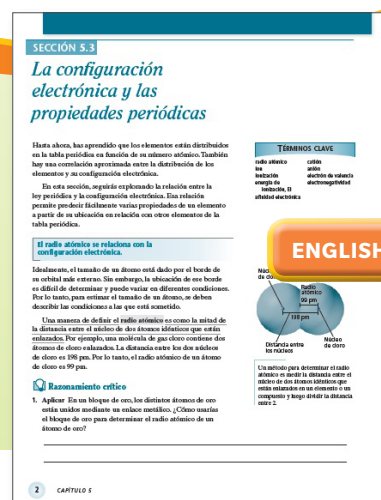
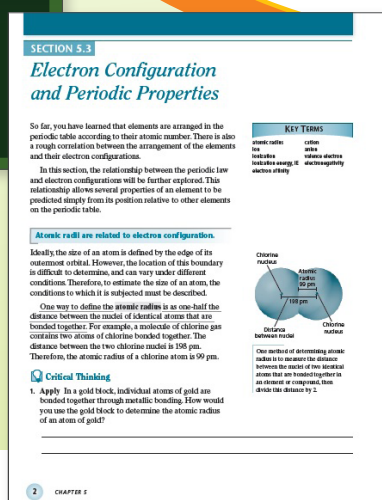
In-text definitions

As students study, they'll find key vocabulary has been highlighted in context.



Interactive Reader

This write-in worktext presents all the vocabulary and essential content from the textbook in a lower-level, easy-to-read text, with instructional visuals and frequent comprehension checks. This unique component is a great tool for all students—the core content for struggling students and a useful study guide for others.

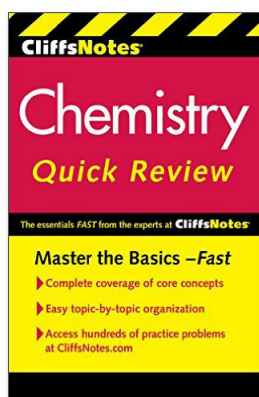


ENGLISH & SPANISH



Interactive Concept Maps

Each chapter includes an interactive, advanced **graphic organizer** that shows the relationships among concepts covered and helps students develop logical thinking and study skills.



CliffsNotes® Chemistry Quick Review

With a Premium package purchase, a class set of these study guides provides **essential reinforcement of core concepts** in an easy-to-use format.

eBook

This **online version** of the print Student Edition features a wealth of **built-in tools** to help students access the content, including the chunking of content around Main Ideas, with frequent comprehension checks, superior support for problem-solving, high-quality instructional visuals, point-of-use references to online animations, Problem-Solving tutorials, and virtual labs that make abstract concepts more concrete. Features include data persistence, on-page media links, bookmarking, search, notes, and highlighting functionality.

Flexible Assessment Tools to Track Student Progress

The comprehensive assessment options located on **HMHSscience.com** bring together all HMH *Modern Chemistry* assessment tools into one convenient place, giving you many choices for the best way to assess your students' learning.



ExamView® Banks

A complete ExamView Software Suite includes all assessment questions for the program and more than **2,300 additional questions** in Bonus Banks.

Name: _____ Class: _____ Date: _____

Assessment
Gases

Section Quiz: Gases and Pressure
In the space provided, write the letter of the term or phrase that best completes each sentence or best answers each question.

1. What causes a gas to exert pressure?
a. collisions
b. density
c. temperature
d. elevation
2. The SI unit for pressure is
a. newton
b. mm Hg
c. pascal
d. liter
3. The pressure exerted by a gas does *not* depend on
a. temperature
b. volume
c. number of moles present
d. the identity of the gas
4. At sea level, the average height of mercury in a barometer is
a. 760 mm
b. 101.325 atm
c. 1.01325 Pa
d. All of the above
5. Standard temperature and pressure are
a. 32°F and 10 atm
b. 0°C and 1 atm
c. 10 K and 1 atm
d. 0°F and 1 atm

ENGLISH & SPANISH

Editable!

Name: _____ Class: _____ Date: _____

Assessment
Chapter Test A

Chapter: Chemical Bonding
In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

1. The charge on an ion is
a. always positive
b. always negative
c. either positive or negative
d. zero
2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron
b. lose two electrons
c. gain one electron
d. gain two electrons
3. If a compound is ionic, it is composed of
a. a positive charged ion
b. the net of
c. The compound
d. Several ions
4. The only pure substance that is a liquid at room temperature is
a. water
b. mercury
c. bromine
d. none of the above

ENGLISH & SPANISH

Editable!

Section Quizzes

A 10-question multiple-choice and short-answer quiz for each section of the textbook. These are designed for student **formative assessment** to aid in remediation.

Chapter Test A & B

Two **full-length** chapter tests include multiple-choice and short-answer questions. Test B is similar to but more challenging than Test A.

Review and Assessment in the Student Edition

The Student Edition contains **multiple levels of assessment** from Formative to Summative along with helpful review questions that assess students' understanding of chapter and section material.

CHAPTER REVIEW

26. An unknown element is shiny and is found to be a good conductor of electricity. What other properties would you predict for it?
a. colorless
b. brittle
c. malleable
d. ductile
27. Use the periodic table to identify the group numbers and general location of the following elements:
a. carbon, C
b. oxygen, O
c. aluminum, Al
d. barium, Ba
28. How can you tell the difference between an element and a compound?
a. Elements are made of only one kind of atom.
b. Compounds are made of two or more different kinds of atoms.
c. Elements are found in nature, but compounds are not.
d. Elements are pure substances, but compounds are not.
29. Write a paragraph that shows that you understand the following terms and the relationship between them: atoms, molecules, compounds, and elements.
30. Pick an object you can see right now. List three of the object's physical properties that you can observe. Can you also observe a chemical property of the object? Explain your answer.
31. **Interpreting Concepts** One way to make chemicals is by combining known substances and adding water to suit your specific needs for a reaction. Explain how this combination is used in a common product or a common process.
32. **Assessing Needs** Look at the table and answer the questions.

Standards-Based Assessment

Record your answers on a separate piece of paper.

MULTIPLE CHOICE

1. A student investigates how adding a substance to water affects the water's boiling point. The data are shown below.

Amount of substance	Boiling point
0 g	100°C
10 g	101°C
20 g	102°C
30 g	103°C

What scientific explanation can the student logically draw from the data she collected?

- A. As water is added, the boiling point decreases.
- B. As water is added, the boiling point increases.
- C. The boiling point of the water solution is always 101°C.
- D. There is a trend of 1°C per gram of substance added.

2. You find an unknown substance that cannot be separated by filtration, evaporation, or distillation. Would the explanation that this substance is an element be scientifically valid?

- A. Yes, because elements cannot be separated by ordinary chemical means.
- B. Yes, because elements are pure substances made of one type of atom.
- C. No, because the substance could be a homogeneous mixture.
- D. No, because the substance could be a compound, which also cannot be separated by physical means.

GRIDDED RESPONSE

3. A student collects the following data about a particular substance.

Substance's Properties	
temperature	125°C
state of matter	solid
mass	89.5 grams
length	2.5 cm
width	2 cm
volume	10.0 cm ³
conductivity	excellent

Based on these data, calculate the substance's density, expressed with the unit g/cm³, for this substance.

SECTION 2 FORMATIVE ASSESSMENT

Reviewing Main Ideas

1. a. What is the main difference between physical properties and chemical properties?
b. Give an example of each.
2. Classify each of the following as either a physical change or a chemical change.
a. tearing a sheet of paper
b. melting a piece of wax
c. burning a log

3. How do you decide whether a sample of matter is a solid, a liquid, or a gas?
4. Contrast mixtures with pure substances.

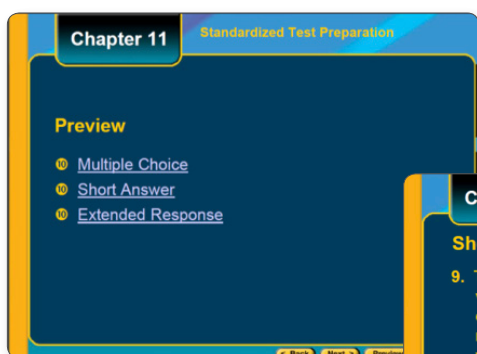
Critical Thinking

5. **ANALYZING INFORMATION** Compare the composition of sucrose purified from sugar cane with the composition of sucrose purified from sugar beets. Explain your answer.



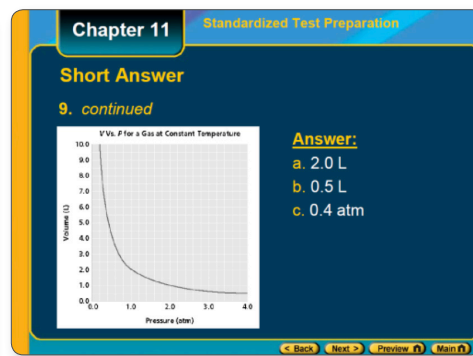
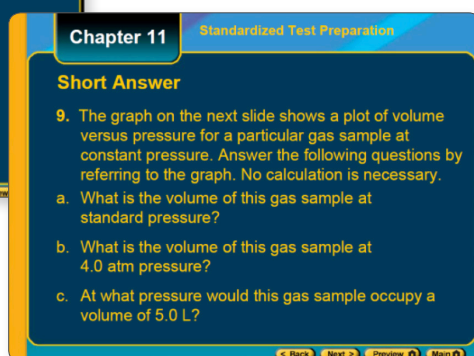
Interactive Review Games

Three different styles of vocabulary and concept review games help reinforce the material learned in each chapter in a fun and engaging format.



PowerPresentations: Standardized Test Preparation

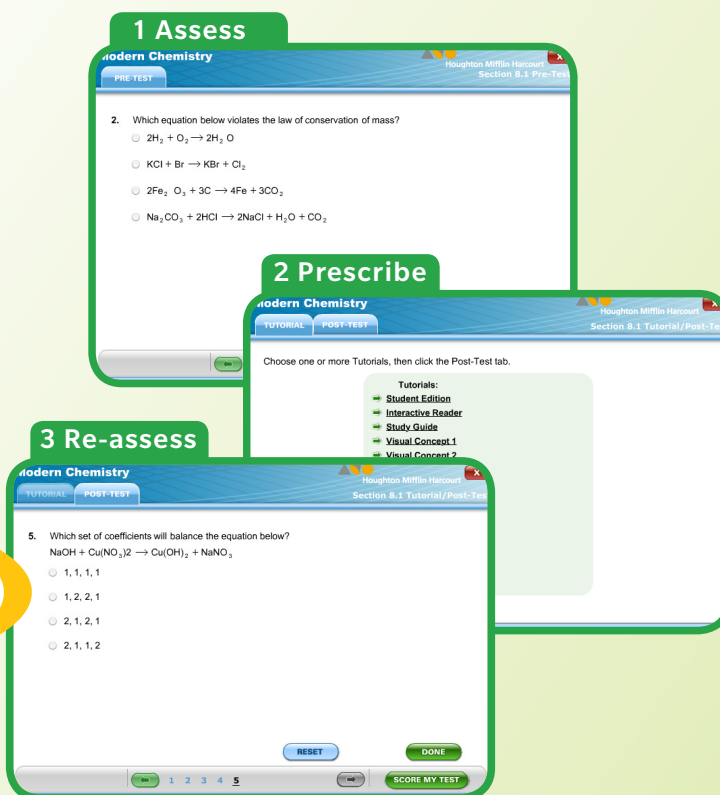
Multiple-Choice, Short-Answer, and Extended-Response questions that you can use for **whole-class review** of chapter materials.



Online Assessment and Remediation

An advanced, automated assessment and remediation engine enables teachers to assign section quizzes to students. The assessments are **automatically graded**, and remediation that uses materials from the program is prescribed. A post-test is offered to determine student mastery. Critical student **performance data** are recorded and made readily available to the teacher.

Individualized remediation



Convenient access to

Labs, Data Analysis and STEM

HMH *Modern Chemistry* includes the most comprehensive lab resources with its wide variety of print and digital lab options for every classroom, along with the most robust data-analysis strand to help students develop these critical skills.

Editable!



Over 200
Editable Labs!

Laboratory Experiments

Wide variety of labs located at point of use on **HMHScience.com**:

- Editable lab sheets
- Teacher notes and answer keys
- Referenced on Instruction and Intervention pages in Teacher Edition

QuickLabs

Designed for reinforcement of key concepts using easy-to-obtain materials.

Standard Labs

Focus on experimental skills and application of chapter concepts through the use of scientific methods.

Core Skill Labs

Provide practice of inquiry skills and scientific methods.

STEM Labs

Science, Technology, Engineering, and Mathematics problem-based labs that emphasize inquiry and the engineering design process.

Open Inquiry Labs

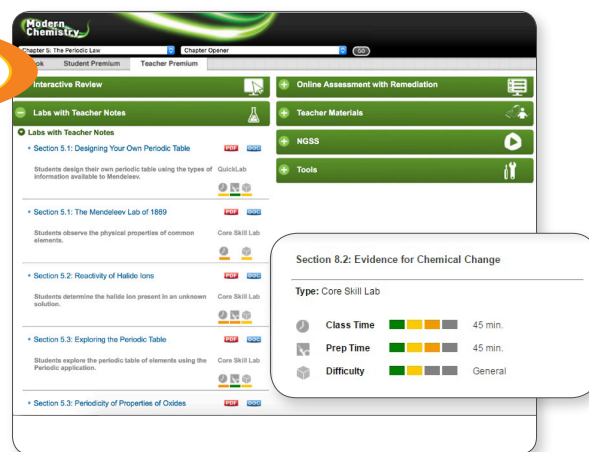
Specifically designed to be short project-based labs that encourage students to collaborate, strategize, construct, and evaluate a lab challenge of their own creation.

Probeware Labs

Labs that use Vernier® probeware and Pasco® probeware and SPARK® technology.

Forensic Labs

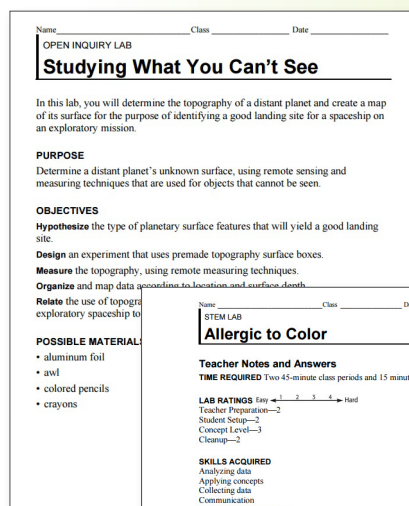
Application labs that have students demonstrating laboratory skills through the exploration of forensic and applied science scenarios.



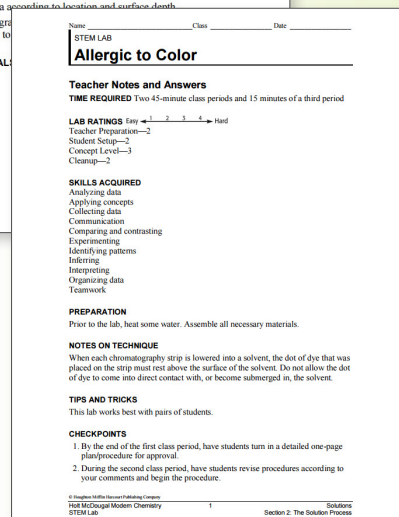
Labeled Labs

Lab activities are labeled online by **class time**, **prep time**, **difficulty** to help teachers choose appropriate activities to fit their classroom needs.

Inquiry labs

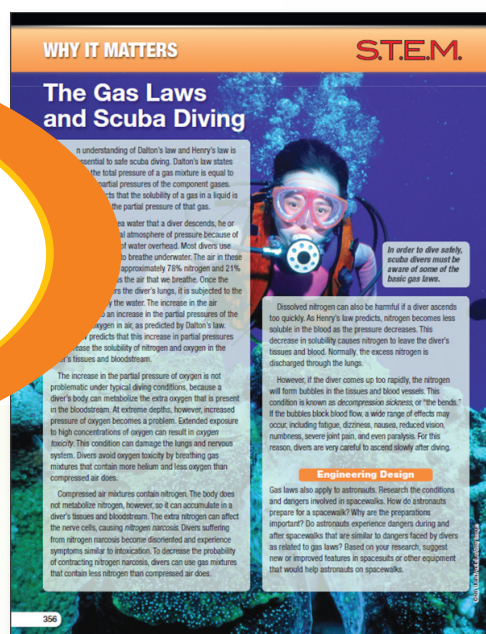


STEM labs



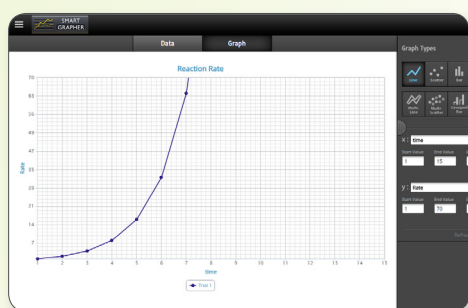
STEM in the Student Edition

Select STEM features now include an Engineering Design feature in the Student Edition. This feature encourages students to follow the engineering design process and think about problems in an innovative way.



Data Analysis Support for Students

To help students develop the data analysis skills necessary to collect, graph, and analyze data like scientists, [HMHSience.com](https://www.hmhsience.com) includes resources to support the data analysis lesson in every chapter.



Smart Grapher

A powerful, easy-to-use **online graphing tool** that encourages students to use their own data to create line graphs, circle graphs, and more.

Scientific Reasoning Skill Builder

Over **100 exercises** that strengthen students' scientific reasoning skills. Sample topics include classifying and categorizing, cause-and-effect relationships, hypothesis, generalizations and analogies, and summarizing and reviewing.

Pre-Lab Procedures

This resource provides **Teacher Resource Pages** and worksheets to help students develop the skills necessary to complete chapter labs.

Graphing Calculator Activities

HMH has partnered with Texas Instruments to present nine **graphing activities** for use with the TI-Nspire™ graphing calculator.



Virtual Labs

Students can hone their lab skills in a virtual environment. Fun, safe and highly interactive, these labs focus on experiments for which equipment and materials are often expensive or difficult to acquire.

The © 2017 **Modern Chemistry** program is available in five configurations: Premium, Hybrid, Digital, Digital Enhanced, and Class Set packages. The Hybrid bundle is the base option, with the print Student Edition and Teacher Edition, the student eBook, and all worksheets, labs, and Spanish resources. The Premium bundle provides added print resources, such as Interactive Readers and CliffsNotes® Study Guides. Digital bundles offer a low-cost, digital-only option. The Premium, Hybrid, Digital Enhanced, and Class Set bundles include the *On the Job* STEM videos and rich multimedia, animations, and simulations. Common Cartridge® options are also available for purchase.

	Student***	Teacher	Print	Digital
Student Edition				
Teacher Edition				
Interactive Reader (and Answer Key for Teacher)** • Online Audio Files (English only)				
Performance Expectations Guide SE/TE				
Engineering Design Guide SE/TE				
CliffsNotes Chemistry Quick Review (with Premium package only)				
Interactive Online Edition				
• NGSS* Correlation Tool				
• Teacher Guide for Google Expeditions				
• Student eBook: Chapter Summaries Audio files and SE pages**				
• Worksheets (Section Study Guides, Chapter Study Guides, Graphing Calculator Activities)				
• Labs** (STEM, Open Inquiry, QuickLabs, Standard, Challenge, Biotechnology, Probeware, Forensic, Virtual Labs)				
• Lab Resources (Labs with Teacher Notes, Laboratory Manager's Professional Reference, Probeware Instruction Sheet, Pre-Lab Procedures, Comprehensive Materials List, Graphing Calculator Instructions)				
• Student Toolkit (Scientific Reasoning Skill Builder, Project Resources, Smart Grapher, FoldNotes, Periodic Table, Glossary, Scientific Calculator, Graphing Calculator)				
• Teacher Toolkit (Teaching Strategies, Classroom Management Resources, Lesson Plans, Project Resources)				
• Multimedia and Activities (Animated Chemistry, Virtual Labs, Why it Matters Videos, Weblinks)				
• Presentation Tools (Teaching Visuals, Interactive Whiteboard Resources, PowerPresentations)				
• <i>On the Job</i> STEM Videos (with Premium package only)				
• Problem Solving Support (Learn It! Videos, Solve It! Cards, Sample Problem Sets, Interactive Demonstrations, Solution Tutor)				
• Interactive Review (Interactive Concept Maps, Interactive Review Games)				
• Online Assessments (ExamView, Section Quizzes, Chapter Tests A&B, Online Assessment with Remediation)				

*Next Generation Science Standards and logo are registered trademarks of Achieve. Neither Achieve nor the lead states and partners that developed the Next Generation Science Standards was involved in the production of, and does not endorse, this product.

Also available in Spanish. *All of the student-facing resources are available to the teacher via the Teacher's Interactive Online Edition.

Contact your HMH Account Executive today to learn more about
Modern Chemistry © 2017: hmhco.force.com/relocator

Apple and the Apple logo are trademarks of Apple Inc., registered in the U.S. and other countries. App Store is a service mark of Apple Inc. TI-Inspire® is a registered trademark of Texas Instruments Incorporated. Common Cartridge® (CC®) is a trademark of IMS Global Learning Consortium, Inc. in the United States and/or other countries. IOS is a registered trademark of Cisco Technology, Inc. and/or its affiliates in the United States and certain other countries. Pre-AP® is a trademark registered and/or owned by the College Board, which was not involved in the production of, and does not endorse, these products. Vernier® is a registered trademark of Vernier Science & Technology. SPARK® and PASCO® are registered trademarks of Pasco Scientific. SMART Notebook™ is a trademark of SMART Technologies ULC in the U.S. and/or other countries. ActivInspire® is a registered trademark of Promethean Limited. Google Cardboard and Google are trademarks or registered trademark of Google, Inc. ExamView® is a registered trademark of Turning Technologies, LLC. Modern Chemistry® is a registered trademark of HMH Publishers LLC. CliffsNotes®, HMH®, and Houghton Mifflin Harcourt™ are trademarks or registered trademarks of Houghton Mifflin Harcourt. © Houghton Mifflin Harcourt. All rights reserved. Printed in the U.S.A. 04/18 WF457963

Connect with us:

